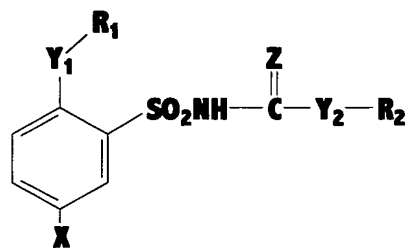


**Benzene-sulphonamide derivatives and their uses****Technical domain**

This invention relates to new benzene-sulphonamide derivatives and to their non-toxic salts as well as to their therapeutic uses.

**Disclosure of the invention**

The new benzene-sulphonamide derivatives, according to the invention, are represented by the general formula (I) :



(I)

in which:

X represents a nitro, cyano, halogen group, eventually radioactive .

Y<sub>1</sub> represents a secondary or tertiary amino group, a sulphur or an oxygen;

Y<sub>2</sub> represents a nitrogen, an oxygen or a -NH group;

Z represents oxygen, sulphur, -N-CN or -CH-NO<sub>2</sub>; and

R<sub>1</sub> and R<sub>2</sub>, which can be identical or different, represent each independently a linear or ramified alkyl group, saturated or unsaturated with 2 to 12 carbon atoms, an alicyclic group, saturated or unsaturated with 3 to 12 carbon atoms, eventually radioactive, an aryl group, substituted or not by one or several alkyl groups in C<sub>1</sub>-C<sub>4</sub>, nitro, cyano, trifluoromethyl, carboxy and halogen, or an arylalkyl group,

or R<sub>1</sub> and/or R<sub>2</sub> form with Y<sub>1</sub> and/or Y<sub>2</sub> a 5 to 7 membered heterocyclic group, saturated or unsaturated chains.

with the exception of derivatives for which X is a nitro group, . Y<sub>1</sub> represents a secondary amine group (-NH-), Y<sub>2</sub> represents a -NH group, Z an oxygen, R<sub>2</sub> an isopropyl and R<sub>1</sub> an element selected in the group comprising (m-toluyl, phenyl and cyclooctyl) and with the exception of N-[  
 5 (2-cyclooctylamino-5-cyanobenzene)sulfonyl]N'-isopropyl urea.

This invention refers also to optical isomers of benzene-sulphonamide derivatives covered by the formula (I) or to salts pharmacologically acceptable of these derivatives

10 This invention refers also to salts of these derivatives, covered by the formula (I), by addition of non-toxic basis, for example to sodium and potassic salts, to salts with an organic acid, as an amino acid such as the lysine, the arginine, for example.

15 When, in the general formula (I), one has an asymmetrical carbon atom (as for example when R<sub>1</sub> and/or R<sub>2</sub> represent an arylalkyl group), the invention refers as well as to pure optical isomers than to the racemic mixture.

Preferred classes of compounds according to the formula (I) are,  
 20 especially, those in which the X represents a nitro, cyano, bromo or iodo group, Y<sub>1</sub> represents a -NH group, Y<sub>2</sub> represents a -NH group or an oxygen atom and R<sub>1</sub> and R<sub>2</sub> represent each independently an ethyl, butyl, tert-butyl propyl, isopropyl, pentyl, hexyl, heptyl, octyl, decyl, amyl, cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, cyclooctyl,  
 25 cyclododecyl, 2-cyclohexenyl, m-toluyl, o-toluyl, p-toluyl, phenyl, allyl, adamantyl, norbornyl, caproyl, 3-carboxyphenyl, 2,3-dimethylphenyl, 2,4-dimethylphenyl, 2,5-dimethylphenyl, 2,6-dimethylphenyl, 3,4-dimethylphenyl, 3,5-dimethylphenyl, 2,4,6-trimethylphenyl, furfuryl, benzyl or 1-phenylethyl group.

Another preferred class of these compounds is that in which  $R_2$  and  $Y_2$  form a homopiperidin group and that in which  $R_1$  and  $Y_1$  form a morpholin or homopiperidin group.

Still another particularly interesting class is that made by radioactive derivatives of the invention, and especially the derivatives in which X represents radioactive iodine, such that the  $^{126}\text{I}$  and its radioactive isotopes  $^{125}\text{I}$  and  $^{131}\text{I}$ , and those in which  $R_1$  represents a saturated alicyclic group or unsaturated group with a tritium hydrogen in positions 2 and /or 3 of the cycle.

As one will see hereinafter in a more detailed way, the derivatives complying with the formula (I) are very useful in the prevention and/or treatment of illnesses involving the thromboxan  $A_2$  at different levels, and especially in the cardio-vascular and blood domains, pulmonary domain, reproduction domain and renal domain. They constitute also an excellent radiolabelled pharmacological tool of the thromboxan  $A_2$  receptors.

The present invention concerns, therefore, also the use of these benzene-sulphonamide derivatives and their salts for drug manufacture for the treatment and/or the prevention of the illnesses involving the thromboxan  $A_2$  as well as as radiolabelled pharmacological tools of the thromboxan  $A_2$  receptors and of the pharmaceutical compositions containing these derivatives, these latter or their salts being used alone or in combination with excipients and/or other therapeutic agents having a similar or different activity.

The active compounds of the invention can be administered, according to the invention, under the form of a pharmaceutical composition, in association with different pharmaceutical excipients and this by oral, parenteral, rectal and topical way.

For the oral administration, one will use pills, granules, tablets, capsules, solutions, syrups, emulsions and suspensions containing classic excipients or additives in clinical pharmacy.

By parenteral way, the salts of active products could be administered in aqueous solution for example.

For the rectal administration, one will use suppositories and, by topical way, lotions, unguents, pomades, aerosols or nebulizers.

5 The active products can be used alone or in combination with other active products having a similar or different activity.

Among the compounds which give, in pharmaceutical use, very interesting results, we have to consider those in the formula (I), in which X represents a NO<sub>2</sub> or iodine group,

10 Y<sub>1</sub> represents a secondary amino group,

Y<sub>2</sub> represents a -NH group,

Z represents an oxygen group, sulphur group or -N-CN group,

and R<sub>1</sub> represents a cyclohexyl group, cycloheptyl group or cychlohexen-2-yl group, and

15 R<sub>2</sub> an isopropyl group, tert-butyl group, pentyl group or homopiperidin group,

and particularly considering the following compounds:

N-[(2-cyclohexylamin-5-nitrobenzene)sulfonyl]N'-tert-butyl urea,

N-cyano-N'-[(2-metatoluy lamin-5-

20 nitrobenzene)sulfonyl]homopiperidinoamidine,

N-[(2-cycloheptylamin-5-nitrobenzène)sulfonyl]N'-cyclohexyl thiourea, and

N-[(cyclohexen-2-yl)-5-iodobenzene)sulfonyl]N'-pentyl urea.

#### Best way to realize the invention

25

Hereafter the definitions and explanations related to the synthesis of the derivatives of the invention are given.

The evolution of most reactions is followed by thin layer chromatography (T.L.C.). The plates are constituted of aluminium foils covered with silica gel 60F<sub>254</sub> (Merck®). The plate is examined by  
30 ultraviolet rays at 254 or 362 nm.

The elementary analysis (C, H, N, S) have been realized and correspond to the theoretic formula (+/-0,4%). The IR and [<sup>1</sup>H]-RMN spectrums are in accordance with the proposed formulas.

5 The elementary analysis (C, H, N, S) have been determined on an Carlo Erba EA 1108 analyzer .

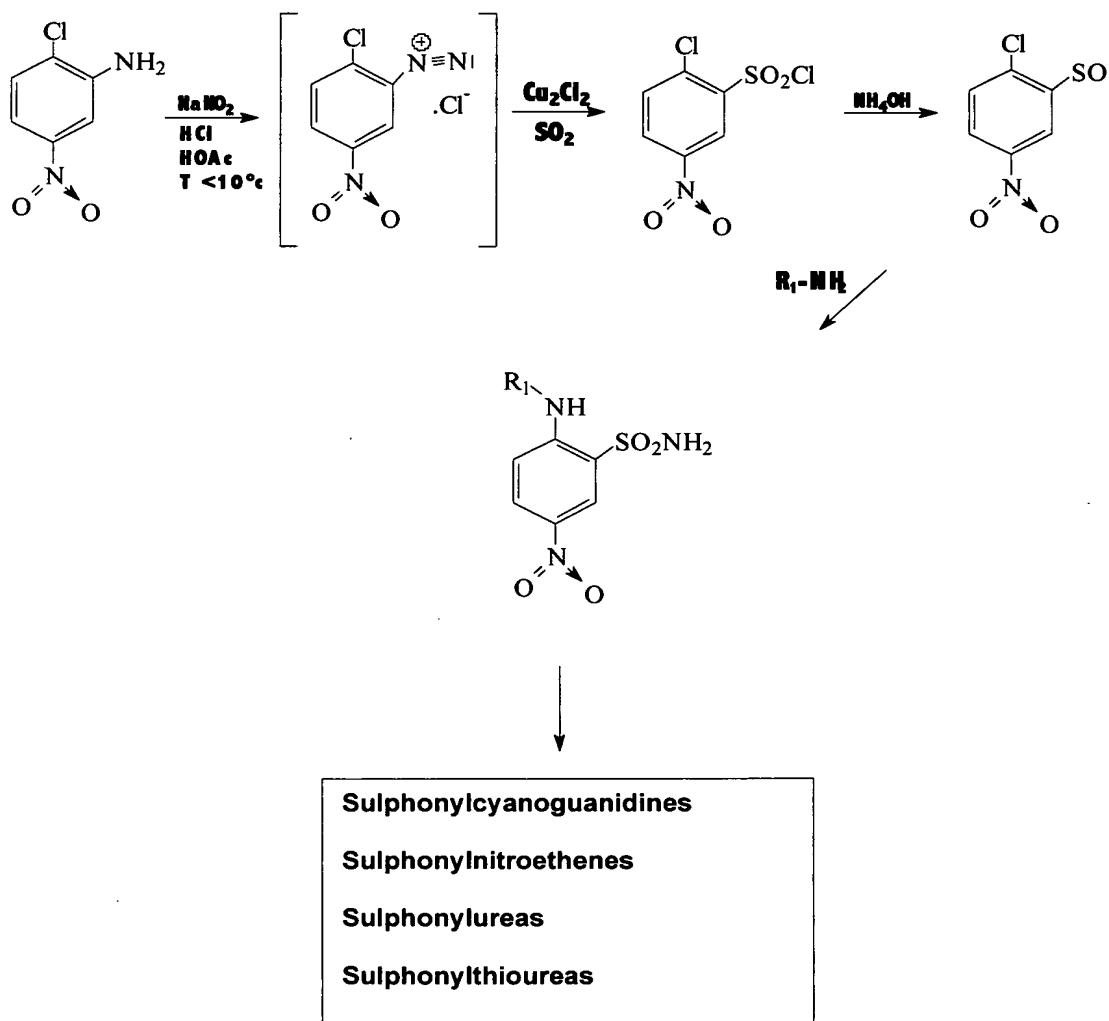
The infrared spectrums of different substances (1 mg) have been registered by means of a FT-IR Perkin-Elmer 1750 under the form of KBr (250 mg) pellets.

10 After dissolution in the deuterium DMSO, the RMN-<sup>1</sup>H spectrum of different molecules is registered on an Bruker 400 apparatus.

The melting points of the obtained molecules have been determined on an Büchi-Tottoli apparatus.

15 The general formula compounds (I) can be obtained easily by different way summarized in the hereafter synthetic schemes.

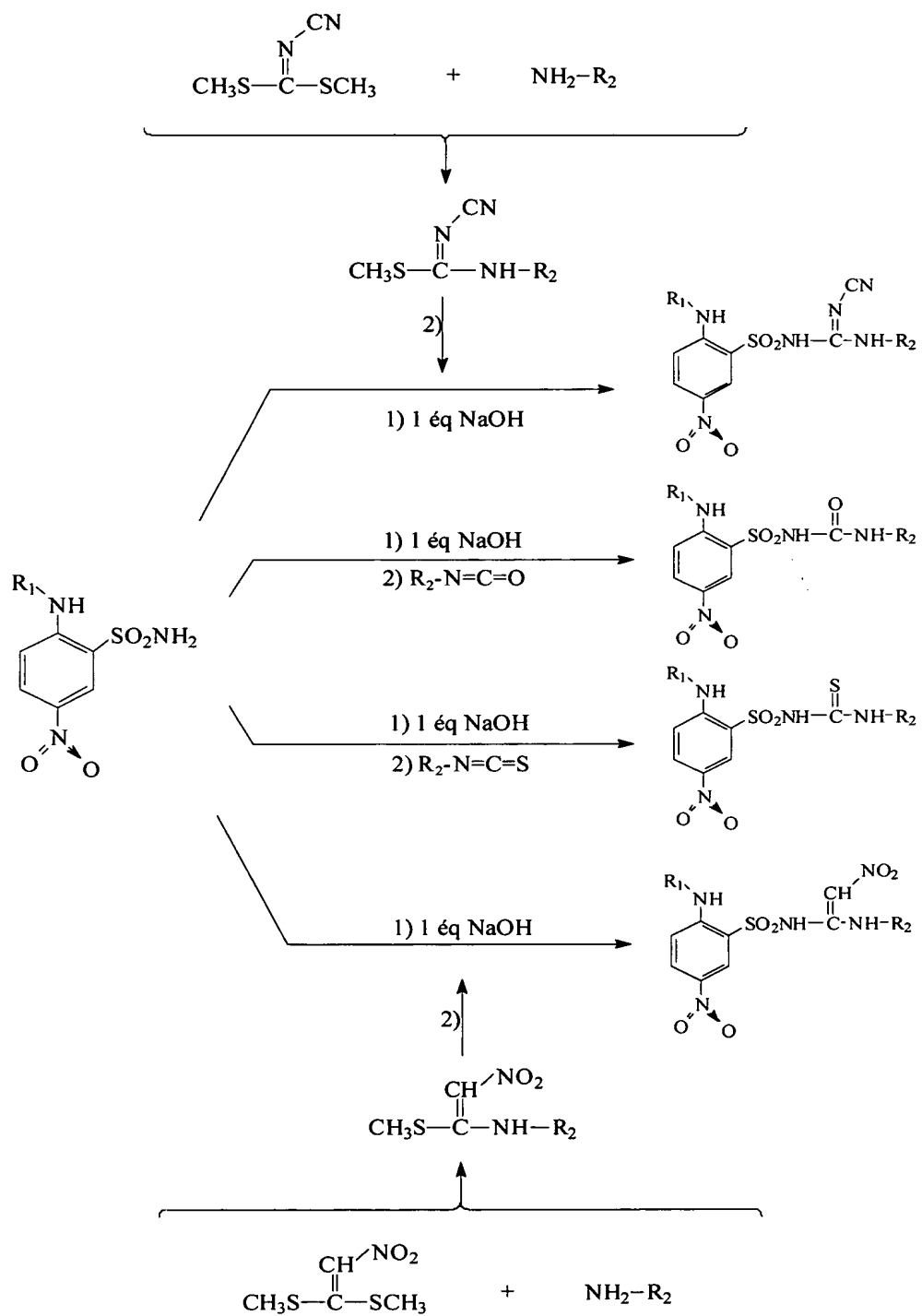
**Scheme 1**  
**Nitrobenzene derivatives**



The 2-chloro-5-nitroaniline is diazotised at a temperature comprised between 0 and 10°C. The diazonium salt formed is substituted in presence of copper salts (catalyst) by sulphur anhydride to generate sulphochloride which in presence of ammonia forms the corresponding 2-chloro-5-nitrobenzenesulfonamide. The chlorine is then substituted by an adequate amine.

The adequated sulphonylurea, thioureas, cyanoguanidines and nitroethenes functions are obtained by condensation of selected reactives (isocyanates for sulphonylureas or isothiocyanates for sulfonylthioureas) or prepared (N-cyano-N'-alkyl (or aryl)carbamimidothioate of S-methyl for sulfonylcyanoguanidines and 1-alkyl (or aryl)amino-1'-methylthio-2-nitroethylene for sulfonitroethenes) on the sulphonamide sodium salt obtained by reaction with exactly 1 sodium hydroxyde equivalent.

## Scheme 1 (following)



### 1.1.) 2-Chloro-5-nitrobenzenesulfonamide

On the one hand, one saturates 160 ml of anhydrous acetic acid in SO<sub>2</sub> for 5 hours (solution A), on the other hand, 10 g of 2-chloro-5-nitroaniline are dissolved in 40 ml of 12 N hydrochloric acid and 100 ml of anhydrous acetic acid (solution B). This solution is cooled till reach a temperature near 0 to -5°C. Finally, one dissolves 7 g of sodium nitrite in 10 ml of water (solution C). The solution C is added drop by drop to solution B to form the diazonium salt. The temperature must be maintained towards -5°C. 4 g of CuCl<sub>2</sub> are dissolved in 10 ml of water (solution D). The solution D is added to solution A and agitated for 2 minutes. A precipitate of Cu<sub>2</sub>Cl<sub>2</sub> appears. The diazonium solution is then prudently and under agitation added to this suspension then 180 g of ice is added in the reaction medium. The precipitate of sulphonyl chloride is rapidly collected on filter, washed with cold water and added under agitation to a previously cooled solution, realized with 50 ml of concentrated ammonia and 100 ml of water. After filtration and clarification with charcoal, the filtrate is concentrated under reduced pressure. The pH is adjusted to 5-6 by 10 N hydrochloric acid. After cooling, the sulphonamide is collected on filter, washed with water and dried. Then it is eventually recrystallized with methanol.

Yield: 50-60%.

Melting point: 178°C

Molecular Weight: 236,62 (C<sub>6</sub>H<sub>5</sub>ClN<sub>2</sub>O<sub>4</sub>S)

### 1.2.) 2-Alkyl (or aryl)amino-5-nitrobenzenesulfonamides

10 g of 2-chloro-5-nitrobenzenesulfonamide prepared in 1.1.) are put in solution in 50 ml of 3-chlorotoluene with 15 ml of amine. One heats about 3 hours, under nitrogen. The reaction is followed by thin layer chromatography. At the term, the solution is filtered, then concentrated under reduced pressure. The residue is retaken by a sodium hydroxyde solution at 2% and purified with charcoal. One brings to pH 1 by 2N

hydrochloric acid. The suspension is extracted three times by 100 ml of diethylic ether. The ether is evaporated under reduced pressure. The residue is retaken by a sodium hydroxyde solution at 2%, then clarified with charcoal and brought to pH 7,5-8 by 5N hydrochloric acid.

5           The precipitate of 2-alkyl (or aryl)amino-5-nitrobenzenesulfonamide is collected on filter, washed and recrystallized with methanol.

### 1.3.) Sulphonylureas

#### N-[(2-alkyl (or aryl)amino-5-nitrobenzene)sulfonyl] N'-alkyl (or aryl) ureas

10           One dissolves 0,01 mole of suitable sulphonamide prepared in 1.2.) in 30 ml of a water-acetone mixture (50/50 vol/vol). After having added a sodium hydroxyde equivalent (solution at 10%), one adds 0,02 mole of adequate isocyanate. For weak volatile isocyanates (B.P. >90°C), the solution is brought to reflux under agitation while for volatile isocyanates

15           (isopropyl-, ethyl-, methylisocyanate), the solution is placed under agitation at room temperature. The progression of the reaction is followed by thin layer chromatography. At the end, the reaction medium is evaporated under depression, the residue is retaken by 100 ml of sodium hydroxyde at 2%. This solution is extracted three times by 150 ml of diethylic ether then clarified with charcoal. The aqueous phase is brought

20           to pH 7,5 by 2N hydrochloric acid. The sulphonylurea which precipitates is collected on filter, washed with water and dried. The product is eventually recrystallized in diluted alcohol.

25           Examples of compounds prepared according to this process (Table 1) :  
n° 1; 2; 13; 17; 19; 20; 21; 22; 23; 25; 26; 27; 28; 29; 30; 31; 32; 33; 34;  
44; 46; 47; 48; 49; 50; 52; 53; 54; 55; 56; 57; 58; 59; 60; 61; 62; 63; 64;  
65; 67; 73; 75; 76; 77; 78; 79; 80; 81; 82; 83; 84; 85; 86; 87; 88; 89; 90;  
91; 92; 94; 95; 96.

30

#### 1.4.) Sulfonylthioureas

##### N-[(2-alkyl (or aryl)amino-5-nitrobenzène)sulfonyl] N'-alkyl (or aryl) thioureas

5           0,01 mole of suitable sulphonamide prepared in 1.2.) is dissolved in 30 ml of a water-acetone mixture (50/50 vol/vol) . After having added a sodium hydroxyde equivalent (solution at 10%), 0,02 mole of adequate isothiocyanate is added. For weak volatile isothiocyanates (B.P.. >90°C), the solution is brought to reflux under agitation while for volatile

10           isothiocyanates (isopropyl, ethyl, methylisothiocyanate), the solution is placed under agitation at room temperature. The progression of the reaction is followed by thin layer chromatography. At the end, the reaction medium is evaporated under depression, the residue is retaken by 100 ml of sodium hydroxyde at 2%. This solution is extracted three times by 150

15           ml of diethylic ether then clarified with charcoal. The aqueous phase is brought to pH 7,5 by 2N hydrochloric acid. The sulfonylthiourea which precipitates is collected on filter, washed with water and dried. The product is eventually recrystallized in diluted alcohol.

20           Examples of compounds prepared according to this process (Table 1) :  
n° 11; 12; 14; 15; 16; 35; 36; 37; 38; 39; 40; 41; 50.

#### 1.5.) Sulfonylcyanoguanidines

##### 1.5.1.) N-cyano-N'-alkyl (or aryl)carbamimidothioates of S-methyl

25           One allows to react 0,05 mole of dimethyl N-cyanodithioiminocarbonate with 0,075 mole of adequate amine in 10 ml of ethanol. This solution is heated under reflux for 15 to 20 hours (for volatile amine, the reaction itself will proceed at room temperature). The progression of the reaction is followed by thin layer chromatography. At

30           the end, the solution is cooled under ice cold water and the precipitate collected on filter, then it is recrystallized into boiling methanol.

1.5.2.) N-[(2-alkyl (or aryl)amino-5-nitrobenzène)sulfonyl] N'-alkyl cyanoguanidines

0,01 mole of suitable sulphonamide prepared in 1.2.) is dissolved in 5 ml of a water-acetone mixture (50/50 vol/vol) and then 0,01 mole of sodium hydroxyde is added (solution at 10%). This solution is placed under agitation for 10 minutes then concentrated under reduced pressure. The residue (sulfonamidate) is solubilized in a mixture constituted of 3 ml of dioxane and 2 ml of dimethylformamide then added with 0,015 mole of adequate S-methyl-N-cyano-N'-alkylcarbamidimidothioate prepared in 1.5.1.). This solution is brought to reflux under agitation. The progression of the reaction is followed by thin layer chromatography. At the end of the reaction, the solution is concentrated under reduced pressure then added with 100 ml of sodium hydroxyde at 2%. This solution is extracted three times by 150 ml of diethylic ether then clarified with charcoal. The aqueous phase is brought to pH 7,5 by hydrochloric acid 2N. The precipitate is collected on filter, washed with water and dried. The product is eventually recrystallized into methanol.

T.L.C. : ethyl acetate 13/cyclohexane 7.

Examples of compounds prepared according to this process (Table 1) :

n° s 3; 4; 5; 6; 7; 8; 9; 18; 51; 74.

1.6.) Sulfonylnitroethenes

1.6.1.) 1-Alkyl (or aryl)amino-1'-methylthio-2-nitroethylnes

One allows to react 0,05 mole of 1,1'-bis(methylthio)-2-nitroethylene with 0,075 mole of adequate amine in 10 ml of ethanol. This solution is brought under reflux 15 to 20 hours (for volatile amine, the reaction itself will proceed to room temperature). The progression of the reaction is followed by thin layer chromatography. At the end, the solution is cooled

under ice cold water and added with 30 ml of water. The obtained precipitate is collected on filter, then recrystallized with boiling methanol.

T.L.C. : ethyl acetate 8/ petroleum ether PE 40/60 12.

5     1.6.2.) 1-Alkyl (or aryl)amino-1'-[2-alkyl (or aryl)amino-5'-nitrobenzenesulfonamide]-2-nitroethylnes

0,01 mole of suitable sulphonamide prepared in 1.2.) is dissolved in 5 ml of a water-acetone mixture (50/50 vol/vol), then 0,01 mole of sodium hydroxyde is added (solution at 10%). This solution is placed under agitation for 10 minutes then concentrated under reduced pressure. The residue (sulfonamidate) is solubilized in a mixture constituted of 3 ml of dioxane and 2 ml of dimethylformamide then added with 0,015 mole of 1-alkyl (or aryl)amino-1'-methylthio-2-nitroethylne adequately prepared in 1.6.1). This solution is brought to reflux under agitation. The progression of the reaction is followed by thin layer chromatography . At the end of the reaction, the solution is concentrated under reduced pressure then added with 100 ml of sodium hydroxyde at 2%. This solution is extracted three times with 150 ml of diethylic ether then clarified with charcoal. The aqueous phase is brought to pH 7,5 by 2N hydrochloric acid. The precipitate is collected on filter, washed with water and dried. The product is eventually recrystallized into methanol.

T.L.C. : ethyl acetate 8/ petroleum ether PE 40/60 12.

Composition example prepared following this process (Table 1) :

25     n° 10.

1.7.) Sulfonylcarbamates

2-Alkyl (or aryl)amino-5-nitrobenzenesulfonylcarbamates of ethyl

0,01 mole of sulphonamide prepared in 1.2.) is dissolved in 10 ml of anhydrous pyridine. Under agitation, drop by drop, a large excess (10 ml) of ethyl chloroformiate is added. The evolution of the synthesis is

followed by thin layer chromatography. At the end of the reaction, about 15 minutes after having added the chloroformate, the solution is evaporated under reduced pressure and the residue retaken by 100 ml of sodium hydroxyde at 2%. After two extractions by 150 ml of diethylic ether, the alkaline solution is clarified with charcoal then neutralized to pH 6,5 with 2N hydrochloric acid. The carbamate precipitate is collected, washed with water and dried under vacuum.

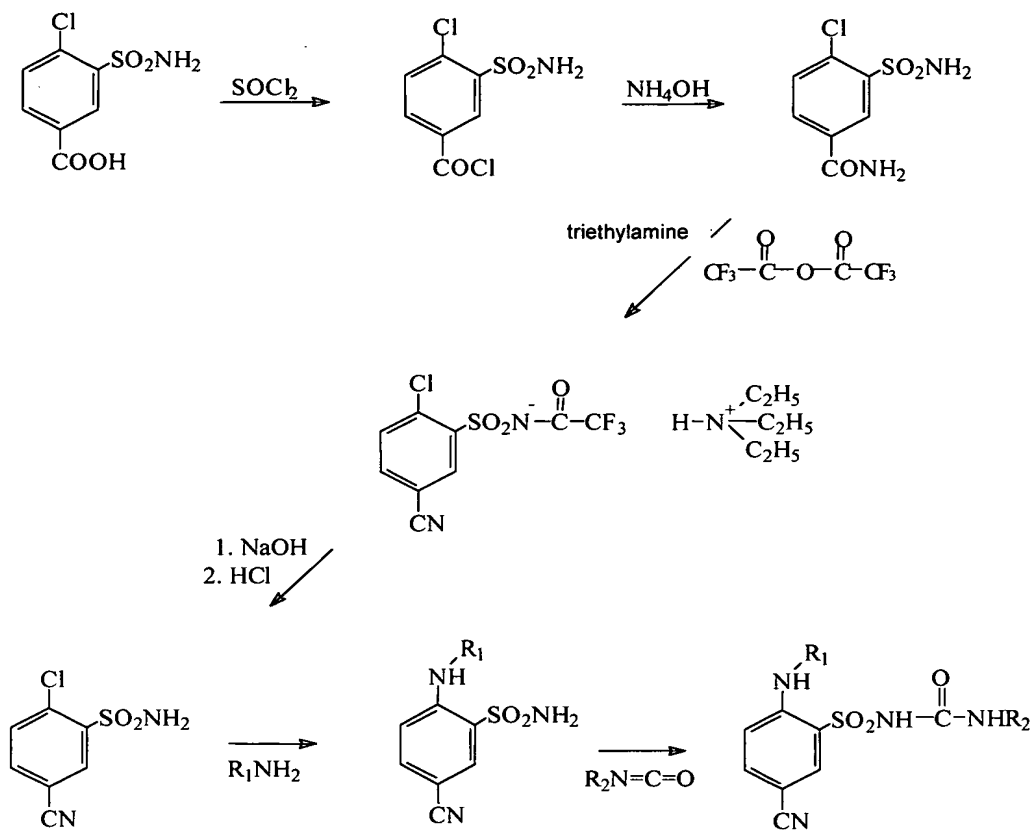
Yield: 75%-88%

T.L.C.: ethyl acetate, methanol and triethylamine 18/2/1.

Example of compounds prepared according to this process (Table 1) : n° 45.

# Scheme 2

## Derivatives of benzonitrile



The 4-chloro-3-sulfamoylbenzoic acid is put in reaction with the thionyl chloride to form the acid chloride which, in presence of ammonia, generates the corresponding carboxamide. The latter is dehydrated in presence of trifluoroacetic anhydride. The acylsulphonamide at the moment of this reaction is hydrolyzed in presence of exactly 2,5 sodium hydroxyde equivalent. The sulphonamide is then regenerated to pH acid. The chlorine is then substituted by an adequate amine. The sulphonylurea function is obtained by condensation of the isocyanate chosen on the previously sulphonamide sodium salt prepared by reaction with exactly 1 sodium hydroxyde equivalent. The carboxylic function is then regenerated by alkaline hydrolysis of the benzonitrile.

#### 2.1.) 4-Chloro-3-sulfamoylbenzenecarboxamide

One allows to react 0,01 mole of acid 4-chloro-3-sulfamoylbenzoic with 25 ml of thionyl chloride. This solution is brought to reflux for 3 hours. At the end, the reaction medium is concentrated under reduced pressure, then added with 10 ml of dioxane. This solution is added under agitation at a previously cooled solution realized with 25 ml of concentrated ammonia and with 50 ml of water. The excess of ammonia is eliminated under reduced pressure. The precipitate is collected on filter, washed with water and dried. It is eventually recrystallized into methanol.

Yield: 50-60%

Point of fusion: 220-222°C

Molecular weight: 234,656 (C<sub>7</sub>H<sub>7</sub>ClN<sub>2</sub>O<sub>3</sub>S).

T.L.C.: ethyl acetate 18/methanol 4/formic acid  
5 drops.

#### 2.2.) 4-Chloro-3-sulfamoylbenzonitrile

To 0,01 mole of 4-chloro-3-sulfamoylbenzenecarboxamide 80 ml of anhydrous tetrahydrofurane are added. This suspension is cooled at 0°C then successively added with 0,045 mole of triethylamine and 0,02 mole of

trifluoroacetic anhydride. The progression of the reaction is followed by thin layer chromatography. At the end, the reaction medium is concentrated under depression. The residue is retaken by water, filtered and washed. The obtained product is put in reaction with 2,5 equivalent of 2N sodium hydroxyde solution for a maximum of 30 minutes. The solution is then brought to pH 1 by 2N hydrochloric acid. The precipitate is then rapidly collected on filter, washed with water and dried.

Yield: 70-80%

Melting point: 199-201°C

10 Molecular weight: 216,64 (C<sub>7</sub>H<sub>5</sub>ClN<sub>2</sub>O<sub>2</sub>S).

Elementary analysis: found: +/- 0,4% of calculated.

T.L.C.: ethyl acetate 18/methanol 4/ formic acid  
5 drops.

### 15 2.3.) 4-Alkyl (or aryl)amino-3-sulfamoylbenzonitriles

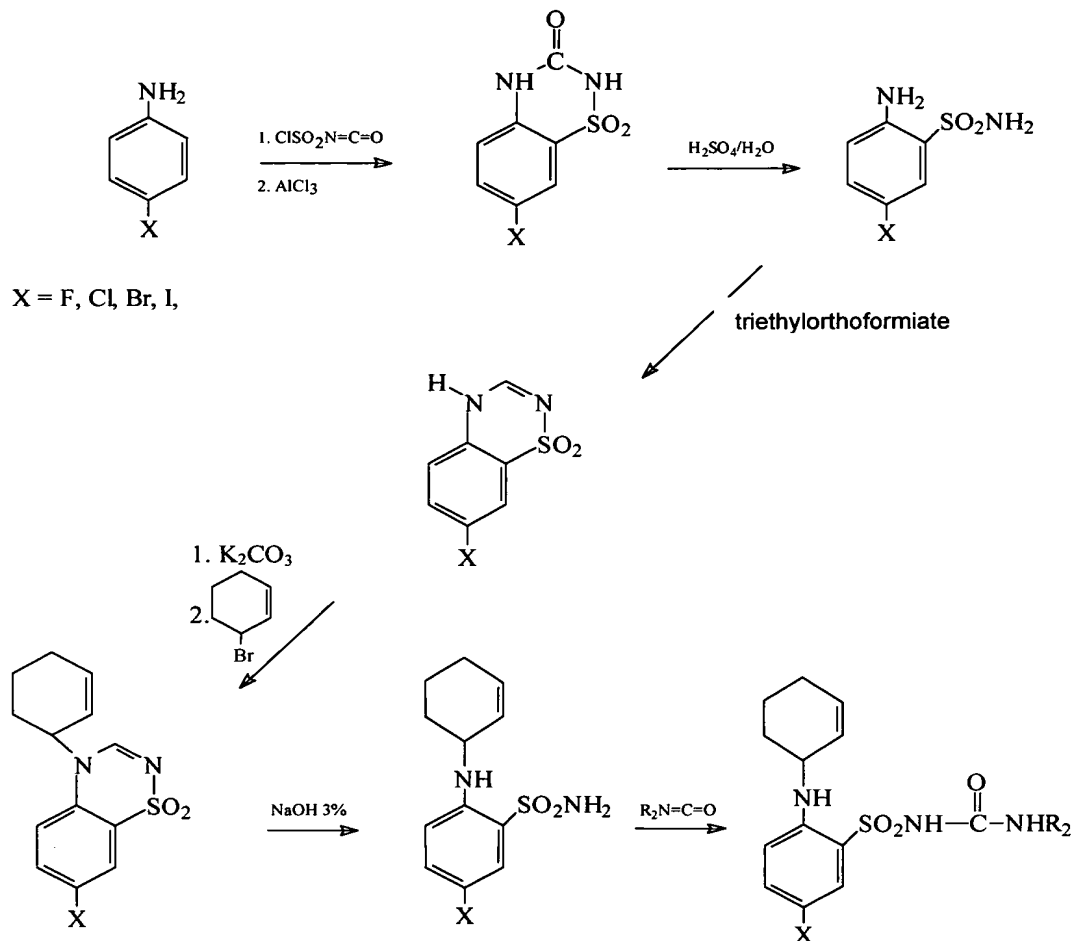
One processes as in 1.2.) by using the 4-chloro-3-sulfamoylbenzonitrile as raw material.

### 20 2.4.) N-[(2-alkyl (or aryl)amino-5-cyanobenzene)sulfonyl] N'-alkyl (or aryl) ureas

One processes as in 1.3.) by using 4-alkyl (or aryl)amino-3-sulfamoylbenzonitrile as raw material.

25 Examples of compounds prepared according to this process (Table 1) : n°  
24; 43; 66; 97.

Scheme 3  
Halogenobenzenic derivatives



5

Halogenobenzenic derivatives

10

The adequate aniline is placed in reaction with a light excess of chlorosulfonylisocyanate at a temperature of  $-5^{\circ}\text{C}$ . Aluminium chloride is then added in the medium in order to obtain the following cyclic product: 2,3-dihydro-7-halogeno-3-oxo-4H-1,2,4-benzothiazidine 1,1-dioxide. The

latter is hydrolyzed by treatment in sulphuric medium. The aminosulfonamide is then engaged in a new reaction of cyclisation to the triethyl orthoformate. The 7-halogeno-4H-1,2,4-benzothiadiazine 1,1-dioxide obtained is alkylated in position 4 with the 3-bromocyclohexene in presence of 4 potassic carbonate equivalent.

The 2-(cyclohexene-2-yl)amino-5-halogenobenzenesulfonamide is then generated by sodium hydroxyde treatment. The sulphonylurea function is obtained by condensation of the chosen isocyanate on the previously sulphonamide sodium salt prepared by reaction with exactly 1 sodium hydroxyde equivalent.

### 3.1.) 2,3-Dihydro-7-halogeno-3-oxo-4H-1,2,4-benzothiadiazine 1,1-dioxides

0,07 mole of chlorosulfonylisocyanate is solubilised in 90 ml of nitromethane previously cooled at -5°C, then drop by drop, 50 ml of a solution of nitromethane containing 0,06 mole of adequate amine, is added. One add drop by drop 0,097 mole of aluminium chloride in the medium. The solution is heated under reflux for 45 minutes, then poured on ice. The obtained precipitate is collected on filter, washed with water and dried. The product is eventually purified by redissolution in a sodium bicarbonate aqueous solution (5% m/vol) and reprecipitation by 2N hydrochloric acid addition.

Yield: 70-75%

T.L.C.: ethyl acetate 20/formic acid 5 drops.

### 3.2.) 2-Amino-5-halogenobenzenesulfonamides

0,01 mole of 2,3-dihydro-7-halogeno-3-oxo-4H-1,2,4-benzothiadiazine 1,1-dioxide prepared in 5.1.) is added to 100 ml of a sulphuric- acid water mixture (50/50). The reaction medium is carried to reflux for one hour. After cooling, the solution is brought to pH 3 by sodium

[illegible]

Yield: 80-85%

T.L.C.: ethyl acetate 13/cyclohexane 7/formic acid  
5 drops.

### 3.3.) 7-Halogeno-4H-1,2,4-benzothiadiazine 1,1- dioxides

5 0,01 mole of 2-amino-5-halogenobenzenesulfonamide prepared in 5.2.) is dissolved in 25 ml of triethylorthoformate. The reaction medium is carried to reflux for one hour. After cooling, the precipitate is collected on filter, washed and dried.

Yield: 50-60%

10 T.L.C.: ethyl acetate 13/cyclohexane 7/formic acid  
5 drops.

### 3.4.) 4-(Cyclohexen-2-yl)-7-halogeno-1,2,4- benzothiadiazine 1,1-dioxydes

15 One puts in suspension 0,01 mole of 7-halogen-4H-1,2,4-benzothiadiazine 1,1-dioxide prepared in 5.3.) in 300 ml of acetonitrile containing 0,04 mole of potassic carbonate. The reaction medium is carried to reflux 30 minutes then added with 0,04 mole of 3-bromocyclohexene. The reflux is maintained during 4 hours. The reaction is followed by thin layer chromatography. At the end, the potassic carbonate in excess is collected on filter. The filtrate is concentrated under reduced pressure. The residue is added with 50 ml of methanol carried to ebullition. The precipitate is collected on filter, washed and dried.

Yield: 60-70%

20 T.L.C.: ethyl acetate 13/cyclohexane 7/formic acid  
25 5 drops.

### 3.5.) 2-(Cyclohexen-2-yl)amino-5-halogenobenzene-sulphonamides

30 To 0,01 mole of 4-(cyclohexen-2-yl)-7-halogeno-1,2,4-benzothiadiazine 1,1-dioxide prepared in 5.4) is added 50 ml of sodium hydroxyde at 3%. The suspension is brought to 60°C for twelve hours. At

the end, the solution is brought to pH 7 by 5 N hydrochloric acid . The obtained precipitate is collected on filter, washed with water and dried.

Yield: 50-60%

T.L.C.: ethyl acetate 13/cyclohexane 7/formic acid

5 5 drops.

3.6.) N-[(2-cyclohexen-2-yl)-5-halogenobenzene)sulfonyl] N'-alkyl (or aryl)urea

10 One processes as in 1.3.) by using 2-(cyclohexen-2-yl)amino-5-halogenobenzenesulfonamide as raw material.

Examples of compounds prepared according to this method (Table 1):

n° 68; 69; 70; 71; 72.

15 The Table 1 given hereinafter refers to preparation of a composed series complying with the general formula (I).

As already specified, the new benzene-sulphonamide derivatives so described are interesting in prevention and/or the treatment of the illnesses involving thromboxan A<sub>2</sub> at different levels and especially:

5

Cardio-vascular and blood diseases:

- Myocardial infarction,
- Thrombus formation and vascular lesions,
- Haemostasis diseases,
- Atherosclerosis,
- Arteriosclerosis,
- Myocardial ischemia,
- Arterial hypertension.

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15

Pulmonary:

- Asthma,
- Bronchospasm,
- Pulmonary hypertension.

20

Of the reproduction:

- Preeclampsia.

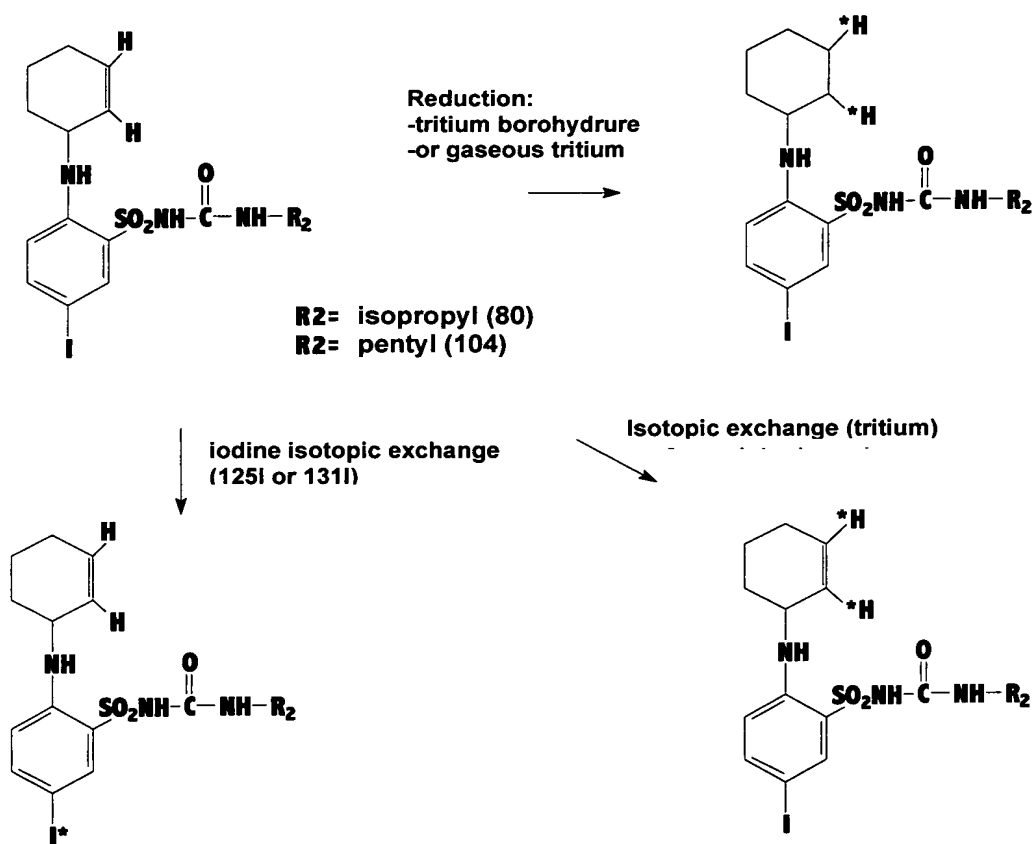
Renal:

- Renal hypertension,
- Renal dysfunction.

25

30

The derivatives of the invention are also interesting for the conception of an original radiolabelled pharmacological tool of thromboxan A<sub>2</sub> receptors. The following scheme 6 shows this kind of application from the compounds n° 80 and 104 (Table 1).



5

As we can see, two labelling technics are considered:

- A tritium marking technique (<sup>3</sup>H).

- either by reduction with a tritium reducer: (tritium hydrogen or tritium borohydride).

10

- either by isotope exchange.

- An iodine labelling technic (<sup>125</sup>I or <sup>131</sup>I) by isotope exchange.

What follows and the tables hereinafter refer to results of pharmacological tests realised on a certain number of compounds given into Table 1.

To operate a first selection, the capacity of these compounds to displace in a specific way a tritium ligand, the [<sup>3</sup>H] SQ-29.548, from the thromboxan A<sub>2</sub> receptor of human platelets have been examined. This binding test is, in fact, simple, fast and allows so a selection of products which have a strong affinity for thromboxan A<sub>2</sub> platelet receptors (TP<sub>α</sub>).

The TXA<sub>2</sub> antagonist potency of the selected compounds has been evaluated by a platelet aggregation test induced by the U-46619 (stable agonist of the thromboxan A<sub>2</sub> ) or by the arachidonic acid.

Two tests on smooth musculature have allowed to confirm the antagonist potency on the TP<sub>τ</sub> thromboxan A<sub>2</sub> receptors. Indeed, the capacity of the selected compounds during the binding to prevent the contraction of the rat fundus induced by l'U-46619 and to relax the rat aorta precontracted by this same stable agonist of the TXA<sub>2</sub> have been evaluated;

All the results are recorded in parallel with those of two thromboxan A<sub>2</sub> receptors antagonists described in literature and which are the object of in-depth clinical studies: the sulotroban and the SQ-29.548.

The SQ-29.548 and the U-46619 are respectively the acid [15-[1- $\alpha$ , 2- $\beta$  (5Z), 3- $\beta$ , 4- $\alpha$ ]-7-[3-[[2-(phenylamino)carbonyl]hydrazin]methyl]-7-oxabicyclo[2.2.1]hept-2-yl]-5-heptenoic and the 9,11-didesoxy,11- $\alpha$ ,9- $\alpha$ -epoxy-methanoprostaglandine F<sub>2a</sub>.

The materials and methods used for pharmacological tests are those described in literature.

**TABLE 2**  
Binding to human platelets thromboxane A<sub>2</sub> receptors  
Binding Test Results on human platelets

COMPOUND NUMBER	BINDING TEST		
	$10^{-6}$ M : (%) <sup>1</sup> AFFINITY	$10^{-7}$ M : (%) <sup>1</sup> AFFINITY	IC 50 <sup>2</sup> ( $\eta$ M)
SULOTROBAN	55,6	16,5	1100
SQ-29.548	100	72,0	23,2
1	93,6	68,0	
2	67,7		
3	20,1		
4	50,0		
5	72,1		
6	29,8		
7	42,9		
8	33,0		
9	15,4		
10	57,7		
11	63,7		
12	67,2		
13	97,7	60,3	
14	92,9	34,0	
15	81,0	16,6	
16	100	46,1	
17	100	88,0	22,7
18	100	88,9	24,2
19	97,8	93,3	3,96
20	1,6		
21	92,2	44,2	
STANDARD DEVIATION <5%			

TABLE 2 (following)

COMPOUND NUMBER	BINDING TEST		
	$10^{-6}$ M : (%) <sup>1</sup> AFFINITY	$10^{-7}$ M : (%) <sup>1</sup> AFFINITY	IC 50 <sup>2</sup> ( $\eta$ M)
22	100	84,1	41,7
23	95,5	62,9	
24	73,7		
25	100	95,2	10,5
26	94,3	93,3	16,9
27	79,6		
28	81,9	39,6	
29	97,4	95,4	7,8
30	95,1	80,8	
31	80,5	42,2	
32	86,7	46,0	
33	86,6	52,4	
34	77,3		
35	45,0		
36	75,6		
37	72,3		
38	77,2		
39	74,5		
40	94,4	63,0	26,9
41	75,9		
42	92,3	50,5	
43	50,0		
44	80,2	51,3	
45	79,9	50,4	
STANDARD DEVIATION <5%			

[illegible][illegible]

TABLE 2 (following)

COMPOUND NUMBER	BINDING TEST		
	$10^{-6}$ M : (%) <sup>1</sup> AFFINITY	$10^{-7}$ M : (%) <sup>1</sup> AFFINITY	IC 50 <sup>2</sup> ( $\eta$ M)
70	63,8		
71	77,8		
72	86,5	52,7	
73	98,3	95,6	1,2
74	90,9		
75	93,1		
76	97,6	93,5	3,5
77	79,4		
78	95,3	71,6	4,2
79	96,6		
80	98,6	97,9	2,4
81	93,3	65,0	57,8
82	98,5	98,0	4,5
83	98,5	92,7	4,5
84	96,9	73,7	23,9
85	92,9	42,5	107,2
86	98,4	94,3	1,83
87	95,6	76,0	18,1
88	95,4	82,0	16,2
89	96,6	83,5	11,5
90	96,9	88,6	5,46
91	97,3	90,8	3,31
92	98,8	95,2	1,62
DEVIATION STANDARD <5%			

TABLE 2 (following)

COMPOUND NUMBER	BINDING TEST		
	$10^{-6}$ M : (%) <sup>1</sup> AFFINITY	$10^{-7}$ M : (%) <sup>1</sup> AFFINITY	IC 50 <sup>2</sup> ( $\eta$ M)
93	97,9	90,2	7,8
94	98,4		2,82
95	98,5		1,45
96	92,3		43,95
97	89,7		98,48
DEVIATION STANDARD <5%			

5

1 Affinity means the per cent of [<sup>3</sup>H]SQ-29.548 specifically substituted by the examined compound.

2 IC 50 : Means the concentrations required for replacing 50% of [<sup>3</sup>H]SQ-29.548 bound to receptors TP $\alpha$ .

10

Test according to :

Cozzi P., Giordani A., Menichincheri M., Pillan A., Pinciroli V., Rossi A., Tonani R., Volpi D., Tamburin M., Ferrario R., Fusar D., Salvati P.,  
- Agents combining thromboxane receptor antagonism with thromboxane synthase inhibition : [[2-(1H-imidazol-1-yl)ethylidene]amino]oxy]alkanoic acids. - *J. Med. Chem.*, **1994**, 37, 3588-3604.

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**TABLE 3 : Platelet Aggregation**  
**Test Results on Human Platelets aggregation**

COMPOUND	AGGREGATION PLATELET TEST	
	ARACHIDONIC ACID IC 50 <sup>1</sup> (μM)	U-46.619 IC 50 <sup>1</sup> (μM)
SULOTROBAN	11,7	10,5
SQ-29.548	0,035	0,034
18	0,36	0,48
STANDARD DEVIATION <5%		

- 5 1IC 50 : Means concentrations required for reduction by 50% the platelet aggregation induced by 0,6 nM of arachidonic acid (AA) or by 30 nM of U-46619.

Test described according to :

- 10 Born G.V.R., Cross M. J., - The aggregation of blood platelets. - *J. Physiol.*, **1963**, 168, 178-195.

Tsuyoshi T., Masayuki Y., Shuichi W., Kazuhiro K., Takashi Y., - Designe, synthesis, and pharmacology of 3-substituted sodium azulene - 1 sulfonates and related compounds : Non-prostaboid thromboxane A<sub>2</sub> receptor antagonists. - *J. Med. Chem.*, **1993**, 36, 791-800.

15

**TABLE 4 : Rat Aorta Contraction**  
**Test results of Rat Aorta Contraction**

COMPOUND	<u>AORTA RAT CONTRACTION TEST</u> IC 50 <sup>1</sup> (ηM)
SULOTROBAN	1,6.10 <sup>3</sup>
SQ-29.548	31,8
17	1,38
18	1,21
22	37,6
25	19,7
29	20,6
40	17,7
STANDARD DEVIATION <5%	

5 1IC 50 : Means the compounds concentrations reducing by 50% the Rat  
Aorta muscular tonus induced by U-46619 (0,03  $\mu$ M).

**Test described according to :**

de Tullio P., Pirotte B., Lebrun P., Fontaine J., Dupont L., Antoine M. H.,  
10 Ouedraogo R., Khelili S., Maggetto C., Masereel B., Diouf O., Podona T.,  
Delarge J., 3-and-4-substituted 4H-pyrido[4,3-e]-1,2,4-thiadiazine 1,1-  
dioxides as potassium channels openers : synthesis pharmacological  
evaluation, and structure-activity relationships. - *J. Med. Chem.*, **1996**, 39,  
937-948.

**TABLE 5 : Rat Fundus Contraction**  
**Test Results for preventing contraction of rat fundus**

COMPOUND	TEST FOR PROVENTING THE <u>RAT FUNDUS CONTRACTION</u> IC 50 <sup>1</sup> (μM)
SULOTROBAN	0,83
SQ-29.548	0,18
18	0,07
STANDARD DEVIATION <5%	

5

1IC 50 : Means compounds concentrations reducing of 50% of maximum contraction amplitude caused by 5 μg de U-46619.

Test description according to :

10

Harris N., Greenberg R., Phillips M. B., Michel I. M., Goldenberg H. J., Haslanger M. F., Steinbacher T.E., - Effects of SQ-27,427, a thromboxane A2 receptor antagonist, in the human platelet and isolated smooth muscle. - *Eur. J. Pharmacol.*, **1984**, 103, 9-18.

15

**TABLE 1**

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
1	1.3.	NO <sub>2</sub>	NH	NH	O	cycloheptyl	isopropyl	153-155	74,6
2	1.3.	NO <sub>2</sub>	NH	NH	O	cyclopentyl	isopropyl	141-143	72,3
3	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	m-toluyI	isopropyl	170-172	62,0
4	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	cyclopentyl	cyclohexyl	172-174	51,5
5	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	cyclohexyl	cyclohexyl	179-181	58,7
6	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	m-toluyI	cyclohexyl	175-177	33,7
7	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	cyclohexyl	isopropyl	168-170	32,2
8	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	cycloheptyl	isopropyl	153-155	46,0
9	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	cyclooctyl	isopropyl	148-150	36,2
10	1.6.2.	NO <sub>2</sub>	NH	NH	CH-NO <sub>2</sub>	m-toluyI	cyclohexyl	176-178	46,5
11	1.4.	NO <sub>2</sub>	NH	NH	S	m-toluyI	isopropyl	134-136	60,8
12	1.4.	NO <sub>2</sub>	NH	NH	S	cycloheptyl	isopropyl	146-148	66,5
13	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	isopropyl	149-151	70,1
14	1.4.	NO <sub>2</sub>	NH	NH	S	cyclohexyl	isopropyl	140-142	34,4
15	1.4.	NO <sub>2</sub>	NH	NH	S	cyclooctyl	isopropyl	160-162	52,5
16	1.4.	NO <sub>2</sub>	NH	NH	S	cyclohexyl	cyclohexyl	167-169	40,8

**TABLE 1 (following)**

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
17	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	cyclohexyl	181-183	50,2
18	1.5.2.	NO <sub>2</sub>	NH	---	N-CN	m-tolyl	[homopiperidine]	161-163	5,4
19	1.3.	NO <sub>2</sub>	NH	NH	O	m-tolyl	tert-butyl	81-83	75,2
20	1.3.	NO <sub>2</sub>	NH	NH	O	propyl	isopropyl	138-140	80,8
21	1.3.	NO <sub>2</sub>	NH	NH	O	benzyl	isopropyl	144-146	74,3
22	1.3.	NO <sub>2</sub>	NH	NH	O	cycloheptyl	cyclohexyl	174-176	48,8
23	1.3.	NO <sub>2</sub>	NH	NH	O	cyclooctyl	cyclohexyl	150-152	45,4
24	2.4.	CN	NH	NH	O	m-tolyl	isopropyl	133-135	28,3
25	1.3.	NO <sub>2</sub>	NH	NH	O	cycloheptyl	tert-butyl	135-137	68,2
26	1.3.	NO <sub>2</sub>	NH	NH	O	cyclooctyl	tert-butyl	136-138	61,3
27	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	ethyl	163-164	72,2
28	1.3.	NO <sub>2</sub>	NH	NH	O	cycloheptyl	ethyl	153-155	74,3
29	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	tert-butyl	147-149	70,2
30	1.3.	NO <sub>2</sub>	NH	NH	O	o-tolyl	isopropyl	109-111	74,3
31	1.3.	NO <sub>2</sub>	NH	NH	O	phenyl	allyl	150-152	53,2
32	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	allyl	152-154	56,3

TABLE 1 (following)

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
33	1.3.	NO <sub>2</sub>	NH	NH	O	cycloheptyl	allyl	138-140	58,2
34	1.3.	NO <sub>2</sub>	NH	NH	O	cyclooctyl	allyl	159-161	47,3
35	1.4.	NO <sub>2</sub>	NH	NH	S	propyl	isopropyl	151-153	72,7
36	1.4.	NO <sub>2</sub>	NH	NH	S	benzyl	isopropyl	149-151	62,8
37	1.4.	NO <sub>2</sub>	NH	NH	S	cyclopentyl	isopropyl	156-158	68,9
38	1.4.	NO <sub>2</sub>	NH	NH	S	cyclohexyl	isopropyl	149-151	63,7
39	1.4.	NO <sub>2</sub>	NH	NH	S	cycloheptyl	ethyl	162-164	62,4
40	1.4.	NO <sub>2</sub>	NH	NH	S	cycloheptyl	cyclohexyl	172-174	38,3
41	1.4.	NO <sub>2</sub>	NH	NH	S	cyclooctyl	cyclohexyl	177-179	30,3
42	1.4.	NO <sub>2</sub>	NH	NH	S	cyclohexyl	furfuryl	168-169	27,2
43	2.4.	CN	NH	NH	O	cyclohexyl	isopropyl	148-150	32,3
44	1.3.	NO <sub>2</sub>	NH	NH	O	cyclooctyl	ethyl	154-155	60,8
45	1.7.	NO <sub>2</sub>	NH	O	O	cyclopentyl	ethyl	147-149	27,4
46	1.3.	NO <sub>2</sub>	NH	NH	O	caproyl	isopropyl	132-134	25,8
47	1.3.	NO <sub>2</sub>	NH	NH	O	adamantyl	tert-butyl	169-171	54,3
48	1.3.	NO <sub>2</sub>	NH	NH	O	cyclododecyl	isopropyl	162-164	50,8

TABLE 1 (following)

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
49	1.3.	NO <sub>2</sub>	NH	NH	O	2,3-dimethylphenyl	isopropyl	146-148	28,3
50	1.3.	NO <sub>2</sub>	NH	NH	O	p-tolyl	isopropyl	132-134	70,8
51	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	m-tolyl	tert-butyl	180-182	25,3
52	1.3.	NO <sub>2</sub>	NH	NH	O	o-tolyl	tert-butyl	90-92	71,4
53	1.3.	NO <sub>2</sub>	NH	NH	O	3-carboxyphenyl	isopropyl	167-169	24,2
54	1.3.	NO <sub>2</sub>	NH	NH	O	norbornyl	isopropyl	177-179	48,3
55	1.3.	NO <sub>2</sub>	NH	NH	O	norbornyl	tert-butyl	111-113	45,4
56	1.3.	NO <sub>2</sub>	NH	NH	O	tert-butyl	isopropyl	165-167	58,3
57	1.3.	NO <sub>2</sub>	NH	NH	O	hexyl	isopropyl	126-128	75,4
58	1.3.	NO <sub>2</sub>	NH	NH	O	adamantyl	isopropyl	179-181	43,8
59	1.3.	NO <sub>2</sub>	NH	NH	O	hexyl	tert-butyl	112-114	72,8
60	1.3.	NO <sub>2</sub>	NH	NH	O	decyl	isopropyl	99-101	58,3
61	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	pentyl	138-140	60,2
62	1.3.	NO <sub>2</sub>	---	NH	O	[morpholine]	isopropyl	183-185	28,3
63	1.3.	NO <sub>2</sub>	---	NH	O	[morpholine]	tert-butyl	172-174	25,4
64	1.3.	NO <sub>2</sub>	---	NH	O	[homopiperidine]	isopropyl	110-112	22,1

TABLE 1 (following)

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
65	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	phenyl	178-180	27,4
66	2.4.	CN	NH	NH	O	norbornyl	isopropyl	149-151	24,7
67	1.3.	NO <sub>2</sub>	NH	NH	O	p-toluy	tert-butyl	126-128	64,3
68	3.6.	NO <sub>2</sub>	NH	NH	O	2-cyclohexenyl	isopropyl	156-158	23,8
69	3.6.	F	NH	NH	O	2-cyclohexenyl	isopropyl	127-129	12,8
70	3.6.	Cl	NH	NH	O	2-cyclohexenyl	isopropyl	132-134	15,3
71	3.6.	Br	NH	NH	O	2-cyclohexenyl	isopropyl	143-145	18,4
72	3.6.	I	NH	NH	O	2-cyclohexenyl	isopropyl	148-150	17,6
73	1.3.	NO <sub>2</sub>	NH	NH	O	2,3-dimethylphenyl	tert-butyl	159-161	24,8
74	1.5.2.	NO <sub>2</sub>	NH	NH	N-CN	cyclohexyl	tert-butyl	192-194	35,8
75	1.3.	NO <sub>2</sub>	NH	NH	O	1-phenylthyl (rac.)	isopropyl	108-110	38,4
76	1.3.	NO <sub>2</sub>	NH	NH	O	1-phenylthyl(rac.)	tert-butyl	146-148	35,2
77	1.3.	NO <sub>2</sub>	NH	NH	O	1-phenylthyl (S)	isopropyl	108-110	28,3
78	1.3.	NO <sub>2</sub>	NH	NH	O	1-phenylthyl (S)	tert-butyl	113-115	25,4
79	1.3.	NO <sub>2</sub>	NH	NH	O	1-phenylthyl (R)	isopropyl	108-110	23,1
80	1.3.	NO <sub>2</sub>	NH	NH	O	1-phenylthyl (R)	tert-butyl	113-115	22,8

TABLE 1 (following)

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
81	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	propyl	137-139	78,8
82	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	butyl	158-160	72,1
83	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	hexyl	115-117	70,8
84	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	heptyl	117-119	76,3
85	1.3.	NO <sub>2</sub>	NH	NH	O	cyclohexyl	octyl	93-95	65,4
86	1.3.	NO <sub>2</sub>	NH	NH	O	2,4,6-trimethylphenyl	isopropyl	170-172	20,8
87	1.3.	NO <sub>2</sub>	NH	NH	O	3,4-dimethylphenyl	isopropyl	149-151	35,4
88	1.3.	NO <sub>2</sub>	NH	NH	O	3,5-dimethylphenyl	isopropyl	147-149	18,8
89	1.3.	NO <sub>2</sub>	NH	NH	O	2,5-dimethylphenyl	isopropyl	148-150	27,3
90	1.3.	NO <sub>2</sub>	NH	NH	O	2,4-dimethylphenyl	isopropyl	162-164	35,4
91	1.3.	NO <sub>2</sub>	NH	NH	O	2,6-dimethylphenyl	isopropyl	148-150	20,2
92	1.3.	NO <sub>2</sub>	NH	NH	O	2,4,6-trimethylphenyl	pentyl	146-148	18,2
93	3.6.	I	NH	NH	O	2-cyclohexenyl	pentyl	148-150	14,3
94	1.3.	NO <sub>2</sub>	NH	NH	O	o-toluy	pentyl	127-129	68,4

TABLE 1 (following)

COMPOUND NUMBER	SYNTHESIS METHOD	X	Y <sub>1</sub>	Y <sub>2</sub>	Z	R <sub>1</sub>	R <sub>2</sub>	PF, °C	YIELDING, (%)
95	1.3.	NO <sub>2</sub>	NH	NH	O	p-toluyI	pentyl	146-148	70,1
96	1.3.	NO <sub>2</sub>	NH	NH	O	m-toluyI	pentyl	129-131	71,2
97	2.4.	CN	NH	NH	O	cyclohexyl	pentyl	144-146	27,8